

REACTIONS OF QUINOLINIUM AND PYRIDINIUM N-IMINES WITH  
CYCLOPENTADIENONE DERIVATIVES

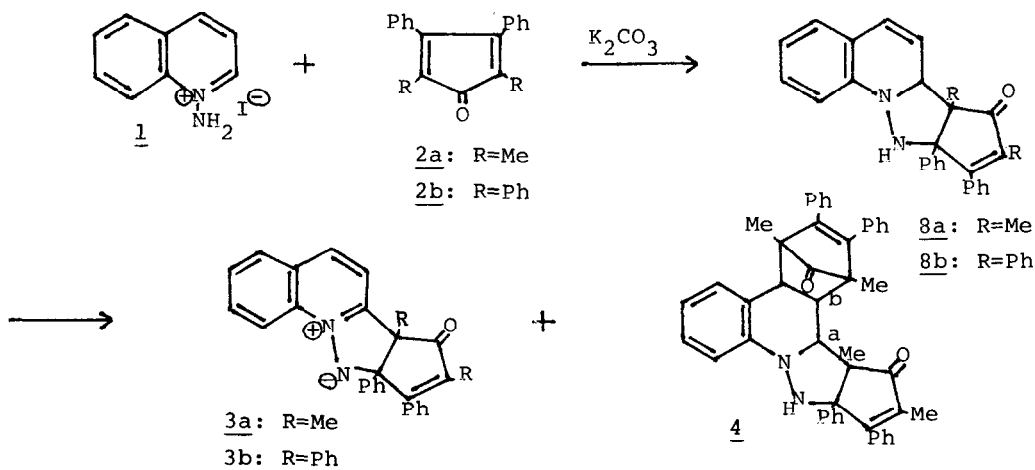
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Summary : The novel reactions of quinolinium and pyridinium N-imines with cyclopentadienones are described.

The 1,3-dipolar addition reactions of pyridinium N-imines and related ylides with activated acetylenes provide a useful method to synthesize nitrogen-bridged heterocycles.<sup>1</sup> In recent years these reactions have been extended to cyclopropanones and yielded some interesting results.<sup>2</sup> For comparisons with cyclopropanones, we have investigated the reactions of quinolinium and pyridinium N-imines with cyclopentadienones and found novel reactions.

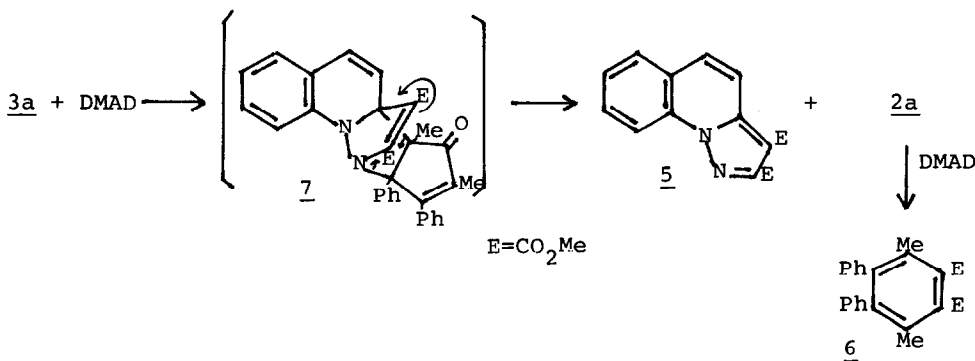
When a mixture of 1-aminoquinolinium iodide (1) and 2,5-dimethyl-3,4-diphenylcyclopentadienone (2a) (1.2:1 equiv) was heated in refluxing acetonitrile in the presence of potassium carbonate for 15 hr, orange crystals 3a (mp 190-192°C, 54%) and colorless crystals 4 (mp 204-205°C, 2%) were obtained.<sup>3</sup> The structures of these products were deduced on the basis of the spectroscopic data along with chemical evidence. 3a;  $\nu_{\max}$  (KBr) 1690  $\text{cm}^{-1}$ ,  $\delta_{\text{ppm}}$  (60MHz,  $\text{CDCl}_3$ )



0.98 (s, 3H), 2.01 (s, 3H), 7.0-7.7 (m, 15H), 8.5-8.8 (m, 1H). 4;  $\nu_{\max}$  (KBr) 3400, 1770, 1695  $\text{cm}^{-1}$ ,  $\delta_{\text{ppm}}$  (60MHz,  $\text{CDCl}_3$ ) 0.59 (s, 3H), 1.22 (s, 3H), 1.35 (s, 3H), 1.48 (s, 3H), 3.09 (s, Ha), 3.25 (d,  $J=9.0$  Hz), 3.50 (d,  $J=9.0$ ), 4.72 (brs, NH), 5.8-6.0 (m, 2H), 6.6-7.8 (m, 22H).<sup>4,5</sup> The elemental analyses, NMR integration, and Mass spectra indicate that 3a is a dehydrogenation product of a 1:1 adduct and 4 is a 1:2 adduct. The ylide structure was suggested by the longer wavelength absorption maximum observed in the UV spectrum of 3a shown in Table 1 as well as no peak due to NH in its IR spectrum. Heating of 3a with dimethyl acetylenedicarboxylate (DMAD) in refluxing toluene gave pyrazolo[1,5-a]quinoline derivative 5 and o-terphenyl derivative 6 in 64% and 60% yields, respectively. The structures of 5 and 6 were established by direct comparisons with authentic samples.<sup>1a, 6</sup> An attractive pathway involves an initial 1,3-dipolar cycloaddition of the ylide 3a with DMAD to give a 1:1 adduct 7 which undergoes a retro-1,3-dipolar addition reaction to afford 5 and 2a. The cyclopentadienone 2a reacts with DMAD to give 6.<sup>6</sup> This reaction confirmed the structure of 3a. The formation of 3a can be explained by dehydrogenation of 1,3-dipolar cycloadduct 8a, while the formation of 4 can be explained by the Diels-Alder reaction of the cyclopentadienone 2a to 8a.<sup>7</sup> Although the intermediate 8a

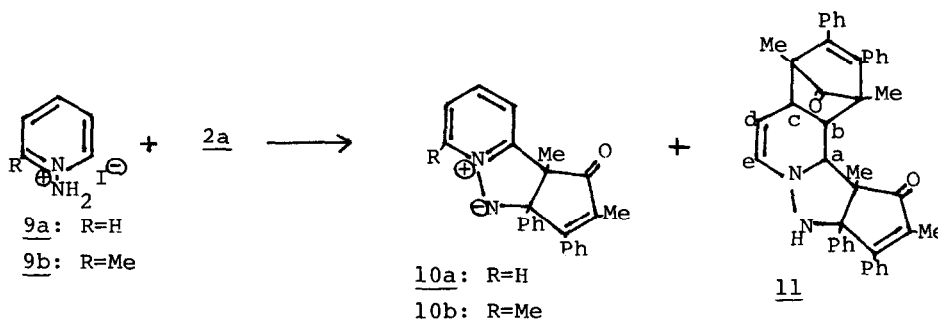
Table 1. Uv spectra of the ylides 3, 10, and the amine 8b

compds	$\lambda_{\max}$	nm, (log $\epsilon$ ) in ethanol			
<u>3a</u>	208(4.58)	225(4.57)	268(4.20)	445(3.51)	
<u>3b</u>	205(4.73)	225(4.70)	278(4.33)	450(3.93)	
<u>10a</u>	206(4.41)	278(4.12)	340(3.96)	430sh(3.45)	
<u>10b</u>	206(4.39)	276(4.13)	330(3.98)	430sh(3.46)	
<u>8b</u>	205(4.70)	231(4.67)	269sh(4.21)	317sh(4.02)	



could not be isolated, the tetraphenyl derivative 8b was isolated in the reaction of the salt 1 with tetracyclone (2b). Upon heating 1 with 2b in refluxing acetonitrile in the presence of potassium carbonate for 30 min, the 1,3-dipolar cycloadduct 8b, mp 189-190°C, was obtained in 74% yield.<sup>3</sup> Treating of 8b with 10% Pd-C in refluxing toluene gave the ylide 3b in 80% yield.<sup>3</sup> The formation of the ylide 3b was confirmed by the comparison of their UV spectra shown in Table 1.

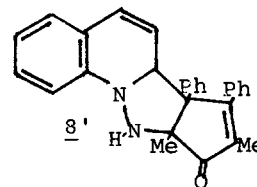
For comparison, the reactions of pyridinium N-imines were investigated. When a mixture of 1-aminopyridinium iodide (9a) and 2,5-dimethyl-3,4-diphenylcyclopentadienone (2a) (1.3:1 equiv) was heated in refluxing acetonitrile in the presence of potassium carbonate for 3 hr, a ylide 10a (mp 201-202°C, 32%) and a 1:2 adduct 11 (mp 185-187°C, 3%) was obtained.<sup>3</sup> Similarly, the reaction of salt 9b with 2a under the same conditions as above afforded the ylide 10b, (mp 213-215°C, 21%) although a 1:2 adduct such as 11 could not be obtained. The structures of these products were determined in a similar manner as those of 3a and 4. 10a;  $\nu_{\max}$  (KBr) 1690  $\text{cm}^{-1}$ ,  $\delta_{\text{ppm}}$  (60 MHz,  $\text{CDCl}_3$ ) 0.90 (s, 3H), 2.05 (s, 3H), 6.55 (ddd, 1H,  $J=7.5, 7.0, 1.5$  Hz), 6.8-7.6 (m, 12H), 7.80 (d, 1H,  $J=7.0$ ). 11;  $\nu_{\max}$  (KBr) 3400, 1760, 1690  $\text{cm}^{-1}$ ,  $\delta_{\text{ppm}}$  (60MHz,  $\text{CDCl}_3$ ) 0.72 (s, 3H), 1.22 (s, 3H), 1.35 (s, 3H), 1.88 (s, 3H), 2.52 (dd, Hc,  $J=9.0, 4.0$  Hz), 3.15 (s, Ha), 3.16 (d, Hb  $J=9.0$ ),<sup>4</sup> 4.56 (dd, Hd,  $J=8.0, 4.0$ ), 4.79 (brs, NH), 6.25 (d, He,  $J=8.0$ ), 7.0-7.8 (m, 20H).<sup>5</sup> The UV spectra of 10 characteristic of the ylide structure are shown in Table 1. The reaction of 10a with DMAD in refluxing toluene gave the o-terphenyl derivative 6 in 32% yield although a pyrazolo-[1,5-a]pyridine derivative corresponding to 5 could not be obtained.



To our best knowledge, the ylides 3 and 10 are first examples of stable N-(alkylimino)quinolinium and pyridinium ylides. This property may be attributed to the fact that the ylides are constructed in a five-membered ring and surrounded by bulky substituents. This point, however, remains equivocal. Further studies are in progress to investigate the reactivities of these novel ylides involving photochemical behaviors.

## REFERENCES AND NOTES

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3. Satisfactory elemental analyses were obtained for all new compounds.
4. The coupling constant between Ha and Hb is nearly zero.
5. The stereochemistry of the 1:2 adducts 4 and 11 is not clear.
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7. The possibility of 1,3-adduct 8' instead of 8 can be denied by the following reason. There are a number of reports concerning the nucleophilic attack of the pyridinium ylides<sup>8</sup> and the reactions of the pyridinium ylides with acrylonitrile<sup>9</sup> and monosubstituted acetylenes<sup>10</sup> give the 1,3-dipolar cycloadducts in which the anion part of the ylides attacks the  $\beta$ -position of the substituted olefins. Furthermore, The methyl signals ( $\delta$  0.98, 2.01) in the nmr spectrum of 3a are located in the lower field compared to those ( $\delta$  0.83, 1.76) of 8a, which was observed when the reaction of 1 with 2a was performed for 1 hr, supporting that 8a is the correct structure because the methyl signals of the ylide formed from 8' can be expected to be in the higher field than those of 8'.
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